

sac.2206

I

II

Ш

KERALA AGRICULTURAL UNIVERSITY

B. Sc. (Hons.) Ag. 2016 Admission

IV Semester Final Examination-July-2018

Problem soils and their management (2+0)

Marks: 50 Time: 2 hours

Fill in the blanks:

(10x1=10)

- ----- is the acidity develops due to hydrogen and aluminium ions concentration of 1 the soil solution.
- 2 ----- is the formula used to calculate potential salinity
- A mineral or organic soil horizon that has a pH less than 3.5, is toxic to plant 3

State True or False

- Acid soils are rich in available calcium and magnesium. 4
- Salt index is negative for good quality water.
- Nitrogen mineralization stops at ammonification stage in submerged soil.

Match the following

 \mathbf{B}

7 Hydrogen sulphide

- Slowly permeable soil a
- Negative logarithm of electron activity 8
- рE b

Compaction technology

Highly permeable soil С

Raised and Sunken bed 10

Akiochi

Write Short notes on ANY FIVE of the following

(5x2=10)

- How does soil pH influence the availability of nutrients and plant growth? 1
- What are the sources of H⁺ ions that make the soil acidic? 2
- Short note on phytoremediation. 3
- 4 How does white alkali soils develop?
- 5 Universal soil loss equation.
- Leaching requirement. 6
- Over-liming of acid soil has adverse effect on nutrient availability and crop growth 7

Answer ANY FIVE of the following

(5x4=20)

- What are all the physical constraints of high clay containing soils? Discuss the method 1 for the management of slowly permeable soils.
- Discuss the causes for the origin of crusted soils and explain the constraints and 2 management practices of crusting soils
- Discuss lime as a component of INM in managing soil health and crop productivity. 3
- Discuss the various agronomic practices for the management of poor quality waters in 4 agriculture
- What is soil conservation and what are the various methods of soil conservation? 5
- Calculate the gypsum requirement to reclaim a sodic soil with an ESP of 40 and CEC of 20 cmol(p⁺)kg⁻¹ to reduce the ESP of upper 30 cm soil to about 10%. Assume the purity of the gypsum to be applied is 70%.

- An irrigation water contains 450, 200, 20, 600, 500 mg L⁻¹ of Na²⁺, Ca²⁺, Mg²⁺, CO₃² HCO₃ respectively. Calculate
 - (a) SAR
 - **RSC** (b)
 - Approximate EC in dS m⁻¹ (c)
 - Osmotic pressure of irrigation water. (d)

IVWrite an essay on ANY ONE of the following

- Nutritional changes (N, P and K) during submergence and sequential reduction process of submerged soil.
- Discuss briefly about the parameters used to assess the quality of irrigation water. Give 2 CSSRI classification of irrigation water.